



## Experimental

Crystals were grown by diffusion of an acetonitrile solution of the title compound into water.

### Crystal data

$C_{31}H_{24}N_4O_8$

$M_r = 580.54$

Orthorhombic

$P2_12_12_1$

$a = 9.849 (2) \text{ \AA}$

$b = 10.800 (2) \text{ \AA}$

$c = 25.682 (3) \text{ \AA}$

$V = 2731.8 (8) \text{ \AA}^3$

$Z = 4$

$D_x = 1.412 \text{ Mg m}^{-3}$

$D_m$  not measured

Cu  $K\alpha$  radiation

$\lambda = 1.54180 \text{ \AA}$

Cell parameters from 25 reflections

$\theta = 5\text{--}22^\circ$

$\mu = 0.868 \text{ mm}^{-1}$

$T = 293 (2) \text{ K}$

Plate

$0.60 \times 0.10 \times 0.05 \text{ mm}$

Colourless

### Data collection

Enraf–Nonius CAD-4

diffractometer

$\omega$ - $2\theta$  scans

Absorption correction: none

3148 measured reflections

3026 independent reflections

2527 reflections with

$I > 2\sigma(I)$

$R_{\text{int}} = 0.016$

$\theta_{\text{max}} = 72.95^\circ$

$h = 0 \rightarrow 11$

$k = 0 \rightarrow 13$

$l = 0 \rightarrow 31$

3 standard reflections

every 100 reflections

intensity decay: none

### Refinement

Refinement on  $F^2$

$R[F^2 > 2\sigma(F^2)] = 0.033$

$wR(F^2) = 0.117$

$S = 0.691$

3026 reflections

413 parameters

H-atom parameters

constrained

$w = 1/[\sigma^2(F_o^2) + (0.1326P)^2 + 0.8103P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.023$

$\Delta\rho_{\text{max}} = 0.164 \text{ e \AA}^{-3}$

$\Delta\rho_{\text{min}} = -0.166 \text{ e \AA}^{-3}$

Extinction correction:

*SHELXL97* (Sheldrick, 1997b)

Extinction coefficient:

0.0032 (4)

Scattering factors from

*International Tables for Crystallography* (Vol. C)

Table 1. Hydrogen-bonding geometry ( $\text{\AA}$ ,  $^\circ$ )

$D\text{---}H\cdots A$	$D\text{---}H$	$H\cdots A$	$D\cdots A$	$D\text{---}H\cdots A$
$N1\text{---}H1\cdots N7^i$	0.86	2.15	2.956 (3)	157
$C2\text{---}H2\cdots O6^i$	0.93	2.56	3.057 (4)	114

Symmetry code: (i)  $\frac{1}{2} + x, \frac{3}{2} - y, -z$ .

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989). Data reduction: *CAD-4 Software*. Program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a). Program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997b). Molecular graphics: *INSIGHTII* (Biosym Technologies, 1995) and *Xtal.GX* (Hall & du Boulay, 1995). Software used to prepare material for publication: *SHELXL97*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: DE1094). Services for accessing these data are described at the back of the journal.

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## 1-Acetyl-5-(4-methoxyphenyl)-3-(4-methoxystyryl)-2-pyrazoline and 1-acetyl-5-(2-chlorophenyl)-3-(2-chlorostyryl)-2-pyrazoline

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## Abstract

The structural details of two methoxy- and chloro-substituted pyrazoline derivatives ( $C_{21}H_{22}N_2O_3$  and  $C_{19}H_{16}Cl_2N_2O$ , respectively) are presented. The two structures show considerable differences in the orientation of the phenyl ring attached to the heterocyclic ring. While the packing of the molecules in the methoxy-substituted derivative is characterized by a  $C\text{---}H\cdots O$  hydrogen bond, the packing in the chloro-substituted derivative is characterized by short  $Cl\cdots Cl$  contacts and  $C\text{---}H\cdots Cl$  intramolecular hydrogen bonds.

## Comment

Pyrazolines are known to exhibit important biological and industrial properties (Wang *et al.*, 1995; El-